

Physical Properties Yahoo Answers

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Physical Properties Yahoo Answers

http://wiki.answers.com/Q/What_is_a_physical_prope... A physical property is a characteristic you can observe or measure without changing the composition. The changes in the physical properties of...

what is physical property? | Yahoo Answers

What physical properties would you use to separate these mixtures? 1) Iron fillings and salt. 2)Salt and water. Name two physical properties that could be used to distinguish between these substances. 3) Water and rubbing alcohol. 4) Gold and aluminum. 5) Helium gas and oxygen gass. Thank you for your help ^_^

Physical Properties? | Yahoo Answers

For the best answers, search on this site <https://shorturl.im/axqfb>. Assuming you mean N₂ - nitrogen gas: Physical properties - Boiling point (77.36 K), Molecular weight (roughly 28 g/mol) Chemical properties - Nitrogen is chemically very stable but it will react with the group I and group II metals to form metal nitrides.

Physical and Chemical properties of Nitrogen? | Yahoo Answers

ACIDS- Substances that produce H⁺ ions in aqueous solution. Physical Properties: 1. Acids have a sour taste eg.lemon juice,vinegar. 2.they have a pH of less than 7

what are physical and chemical properties ... - Yahoo Answers

The corrosion resistance of gold is perhaps one its most useful properties. Electrode potentials are a useful method for representing the tendency of a metal to corrode. Electrode potentials are measured with reference to hydrogen and an electrochemical series can be prepared for metals as indicated below.

what are 6 common physical properties of gold? | Yahoo Answers

Physical property's are,Mass,density,Boiling point,freezing point,Colour,Crystalline,amorphous,being liquid,solid or gas,surface tension,viscosity,elasticity,optical properties etc.

what is physical property and chemical ... - Yahoo Answers

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The chemical and physical properties of gasoline are highly variable depending on the specific product. As well, the hazards of gasoline are affected by the proportion of individual components. For example, gasoline containing a significant proportion of n-hexane may have toxic effects attributable to n-hexane.

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